ChemComm

COMMUNICATION



View Article Online View Journal | View Issue

Cite this: Chem. Commun., 2014, 50, 8558

Received 19th April 2014, Accepted 16th May 2014

DOI: 10.1039/c4cc02910k

www.rsc.org/chemcomm

A bilayer triangular lattice with crown-like Co₇ spin cluster SBUs exhibiting high spin frustration[†]

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A novel bilayer metal–organic framework is assembled with a perfect intralayer triangular subnet and ideal interlayer T_d arrangement between unprecedented crown-like Co₇ cluster units, exhibiting high spin frustration.

Recently geometrically frustrated antiferromagnets have been paid considerable attention in solid state science, due to their great correlation with ground-state behaviors such as spin liquids, spin ices and spin glasses.¹⁻³ Geometric spin frustration occurs only when all nearest neighbor interactions can not be satisfied simultaneously, for example in a triangular or tetrahedral magnetic unit,⁴ which is usually observed in individual or mixed corner- and/or edge-sharing magnetic lattices such as kagomé and triangular lattices in extended dimensional systems.^{5,6} A lot of investigations have centered on the jarosite family of minerals.⁷ Some frustrated magnets have also been reported on the basis of metal-organic frameworks containing cluster units with odd nuclearity, such as trinuclear or pentanuclear clusters with antiferromagnetic couplings.8 Among these, the examples are comparatively common with triangular cluster units $M_3(\mu_3-X)$ (M = transition metal ion, X = OH, O or F) positioned at kagomé or triangular lattice points and bridged by ligands.9 Also, some zero-dimensional materials with similar topological features as kagomé and triangular lattices exhibit geometrical frustration, for example Mn^{III}₇, Fe^{III}₇ disc-like clusters and Fe^{III}₁₂ spin cluster, *etc.*,¹⁰ which are expected to provide a deeper understanding of basic aspects of geometrical frustration.

Herein, we have achieved one highly frustrated 2D compound $\{[Co_7(OH)_6(1,4\text{-npa})_4(H_2O)_3](dmt)_{0.5}\cdot 4H_2O\}_n$ (1) by hydrothermal reaction of $CoCl_2\cdot 6H_2O$, 1,4-npa and dmt (1,4-npa = 1,4-naphthalic acid, dmt = 2,4-diamine-6-methyl-triazine) (ESI†). The chemical formula is determined by elemental analysis. The structure is based on an unprecedented crown-like heptanuclear $Co_7(OH)_6$ spin cluster, which is firstly used as the SBU (secondary building unit) to display a bilayer 2D triangular lattice with π -conjugated 1,4-npa bridges.

Compound 1 crystallized to trigonal space group $R\bar{3}$ (ESI⁺), and shows a 2D lattice consisting of {Co7(OH)6} clusters connected by 1,4-npa. As shown in Fig. 1a, it is observed that there are three crystallographically distinct Co sites within the heptanuclear cluster. All cobalt atoms are assigned as divalent cations, and μ_3 -O and μ_4 -O atoms as hydroxyl oxygen atoms according to the charge balance and the BVS (bond valence sums) calculations.¹¹ On the Co(1) site, one μ_3 -OH, two μ_4 -OH and three carboxyl-O atoms surround the Co center in a slightly distorted octahedron. The Co(2) atom also bears an octahedral environment formed of one μ_4 -OH, two μ_3 -OH, two carboxyl-O atoms and one water molecule. A threefold axis passes through the Co(3) site, the Co(3) atom is therefore ligated by three μ_4 -OH and three carboxyl-O atoms. Three μ_3 -OH and three μ_4 -OH link Co(1), Co(2), Co(3) and their equivalents (Co(1A), Co(1B), Co(2A) and Co(2B)) (A: -y, -2 + x - y, z; B: 2 - x + y, -x, z) in a crown-like heptanuclear cluster with the nearest Co---Co distance of about 2.9 Å (Fig. 1b). As far as we know, reported {Co₇} complexes also



Fig. 1 (a) The coordination environment of the metal ions in **1**. (b) Crownlike $Co_7(OH)_6$ SBU of **1**. H atoms are omitted for clarity. Symmetry codes: A: -y, -2 + x - y, z; B: 2 - x + y, -x, z. Color scheme: C, grey; Co, green; and O, orange.

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[†] Electronic supplementary information (ESI) available: Experimental, crystallographical and physical characterization sections, additional graphics (Fig. S1–S7) and tables for 1 (Tables S1–S3). CCDC 982028. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c4cc02910k

contain a disc-like or trigonal-prismatic unit,¹² but a crown-like heptanuclear cluster is unprecedented. Hydroxide groups bridging in a μ_4 manner are displaced by 0.19 Å out of the best mean plane described by four Co^{II} atoms, which is unusual and few complexes involving transition metal ions are known.¹³ Considering the Co–O–Co angles whose values dictate the sign of the magnetic exchange coupling, it is established that the exchange coupling is antiferromagnetic for angles higher than 100° and ferromagnetic for lower values.¹⁴ For the present case, the Co(2)–O(1)–Co(2), Co(3)–O(2)–Co(2) and Co(1)–O(2)–Co(1) angles of 141.6(2)°, 144.0(1)° and 151.2(2)° can be related to antiferromagnetic coupling, as well as *syn,syn*- $\mu_2:\eta^1:\eta^1$ carboxyl groups.

Each $\{Co_7(OH)_6\}$ cluster as one SBU first connects to the other six adjacent SBUs by six linear 1,4-npa ligands to form one perfect 2D triangular sheet in the ab plane (Fig. 2a), and then additionally links three Co₇(OH)₆ by three other 1,4-npa to form a double layer triangular network. Between the double layer, the octahedral cages are formed by six Co₇(OH)₆ SBUs at the corners, and twelve 1,4-npa ligands at the edges, as well as the ideal tetrahedral cages defined by four Co₇(OH)₆ SBUs at the corners, and six 1,4-npa ligands at the edges (Fig. 2b). The O_b-cage aperture is approximately 5 Å and the open window of about 4×4 Å, and T_d-cage aperture of about 3 Å. Every Oh-cage is therefore surrounded by six Td-cages, and the 2D network may be alternatively described by the formation of O_b-cage and T_d-cage shared corners. From topological views, considering every Co₇(OH)₆ SBU as 9-connected node, and the ligand 1,4-npa as linker, the double layer 2D triangular frustrated framework may be also simplified as a uninodal nine-connected 2D network with Schläfli symbol of $\{3^{15}, 4^{18}, 5^3\}$ (Fig. 2c). The perfect intralayer



Fig. 2 (a) 2D triangular sheet of **1** viewed along the *c*-axis direction. (b) Octahedral and tetrahedral cages. (c) Double layer 2D triangular network of **1** viewed along the *c*-axis direction. Green polyhedrons represent $Co_7(OH)_6$ clusters. Blue and purple balls highlight voids inside cages and H atoms are omitted for clarity. Color scheme: C, grey; Co, green and O, orange.



Fig. 3 The $\chi_m T$ vs. T and ${\chi_m}^{-1}$ vs. T plots of **1** in the range of 2–300 K at 1 kOe; the red solid line is the best-fit; inset: the χ_m vs. T plots of **1** in the range 2–50 K in different fields.

triangular lattice and ideal interlayer tetrahedral arrangement indicate geometrical frustration in 1.⁴ The experimental and simulated PXRD patterns agree well with each other (Fig. S2, ESI[†]), confirming the good phase purity. The thermogravimetric curve (Fig. S3, ESI[†]) shows high stability of compound 1.

The perfect triangular and tetrahedral frustrated connections of 1 drive us to explore the intermolecular magnetic couplings which might be propagated through the extended π -conjugated 1,4-npa bridges between heptanuclear CoII clusters.15 Fig. 3 shows the $\chi_{\rm m}T$ vs. T and $\chi_{\rm m}^{-1}$ vs. T plots of 1 in the range of 2-300 K at 1 kOe. The $\chi_m T$ value is 15.16 cm³ mol⁻¹ K at 300 K, lower than the spin-only value of seven high-spin non-interacting Co^{II} ions at 20.5 cm³ mol⁻¹ K assuming $S_i = 3/2$ and an average g value of 2.5.¹⁶ With the temperature lowered, the $\chi_m T$ value gradually decreases to reach the value of 1.07 cm³ mol⁻¹ K at 2 K, which shows antiferromagnetic behaviour. Unfortunately, no suitable model can be used to fit the magnetic data due to the complicated magnetic couplings in the 2D heptanuclear Co^{II} system of 1. Fitting the temperature dependence of the reciprocal susceptibility (χ_m^{-1}) using the Curie-Weiss law in the temperature 300-25 K, yields large negative θ value (θ = -124.45 K) and C = 21.28 cm³ mol⁻¹ K, which also indicates strong antiferromagnetic interactions.

In order to check for magnetic behaviour at low temperatures, field-cooled magnetizations (FCM) and zero-field-cooled magnetizations (ZFCM) were measured at 100 Oe and 200 Oe (Fig. S4, ESI†). It is observed that the curves are identical and no divergence occurs for ZFCM and FCM curves in the same field, indicating there is no spontaneous magnetization down to 2 K. With the increasing external field, the curves of χ_m at 2–30 K under high field (1–80 kOe) show field-dependent decrease, but no sharp transition indicative of magnetic order appears, even down to 2 K, suggesting no antiferromagnetic long-range ordering occurs and the compound retains its traditional paramagnetic state above 2 K ($T_N < 2$ K) (Fig. 3 inset).¹⁷

The field dependence of the isothermal magnetization M(H) at 2 K was also measured (Fig. S5, ESI[†]), which does not show hysteresis but shows an increase with the field and reaches 2.66 $N\beta$ at 80 kOe, a value that is far below the saturation value 16.8 $N\beta$ expected for seven spin-only Co^{II} ions, which further confirms the antiferromagnetic interactions at low temperature. In addition, the ac susceptibilities in the range of 500–3000 Hz also indicate the above results. The $\chi' vs. T$ curves in a field of 3 Oe are analogous so that no peaks above 2 K and no frequency dependence are observed (Fig. S6, ESI[†]), similar to the reported literature.^{17c}

The observed strong antiferromagnetism may be understood by intramolecular and intermolecular magnetic exchange pathways. Within heptanuclear Co^{II} units, although the angles Co(1)-O(2)-Co(3) $(84.8(1)^{\circ})$, Co(1)-O(2)-Co(2) $(86.4(1)^{\circ})$ and Co(2)-O(1)-Co(1) $(94.1(1)^{\circ})$ define ferromagnetic interactions, Co(2)-O(1)-Co(2) (141.6(2)°), Co(3)-O(2)-Co(2) (144.0(0)°) and Co(1)-O(2)-Co(1) (151.2(2)°) indicate more strong antiferromagnetic interactions,¹⁴ as well as syn,syn-carboxylate bridges serving the antiferromagnetic interactions.¹⁸ Meanwhile, the π -conjugated 1,4-npa bridge provides intermolecular antiferromagnetic coupling between Co7 clusters.¹⁵ The magnetic data also suggest an important spin-frustrated behaviour of 1. Ramirez has provided a measure for spin frustration by defining $f = |\theta/T_{\rm N}|$ with value of f > 10 signifying a strong effect.^{4a,19} According to the definition, 1 shows high spin frustration, with the value of f > 63.2. As far as we know, only a few cases such as [Fe₃(µ₃-O)(µ-OAc)₆(H₂O)₃][Fe₃(µ₃-O)(µ-OAc)_{7.5}]₂·7H₂O and ZnCu₃(OH)₆Cl₂ bear a large f value.^{8b,17a} However, the f values cannot be compared directly between the compounds of isotropic ions such as Fe^{III} and anisotropic Co^{II} ions. The high frustration for **1** may be interpreted as below: (a) 1 has odd heptanuclear $Co_7(OH)_6$ cluster units. The Co_7 cluster can be considered to be a polyhedron consisting of eleven triangular faces (Fig. S7, ESI⁺), of which the frustrated magnetic interactions are dominated by the arrangement of spins on the triangular faces; (b) the trigonal space group $R\bar{3}$ brings the double layer structure of 1 with both its intralayer triangular subnet and its inter-layer T_d arrangement between the Co₇ cluster SBUs perfectly frustrated.

In conclusion, we first report a novel bilayer triangular lattice with unique crown-like $\text{Co}_7(\text{OH})_6$ cluster SBUs, which contain a perfect intralayer triangular network and ideal inter-layer T_d arrangement. Compound **1** presents the first example of the bilayer triangular lattice with high frustration. It offers an opportunity for spin frustrated realization on a molecular level, and provides a deeper understanding of basic aspects of geometrical frustration in extended π -conjugated systems.

The authors gratefully acknowledge support from the Recruitment Program of Global Young Experts, Program for New Century Excellent Talents in University (NCET-10-0108), the Natural Science Foundation of China (21271025, 21273033), the Award 'MinJiang Scholar Program' in Fujian Province, the Department of Science and Technology of Henan Province (122102210174), the Natural Science Foundation of the Education Department of Henan Province (2011A150004, 12B150004) and the State Key Laboratory of Structural Chemistry (2011008). Great thanks also go to P.-T. Ma, L.-S. Wang, X.-D. Zhang and R.-B. Fu for their help.

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